

# Humidity at a Glance

## Most Relevant Equations with Sample Code

This summary provides an overview on the most-used humidity-related formulas. The sample code is optimized for microprocessors (e.g. the common logarithm "log10" is used rather than the natural logarithm "ln"). For an in-depth study of the equations please refer to our complimentary paper "Introduction to Humidity" available on [www.sensirion.com/humidity](http://www.sensirion.com/humidity).

### 1 Relative humidity

Condition: Constant absolute humidity (e.g. closed systems).

$$RH_2 = RH_1 \exp \left[ m \cdot T_n \frac{t_1 - t_2}{(T_n + t_1)(T_n + t_2)} \right]$$

$RH_1$	relative humidity at position 1
$RH_2$	relative humidity at position 2
$t_1$	temperature in °C at position 1
$t_2$	temperature in °C at position 2
$m$	17.62
$T_n$	243.12 °C

Sample code: `RH2 = RH1*exp(4283.78*(t1-t2)/(243.12+t1)/(243.12+t2));`

### 2 Dew point

Definition: The dew point is the temperature to which a given parcel of air must be cooled, at constant barometric pressure, for water vapor to condense into water.

$$t_d(t, RH) = T_n \cdot \frac{\ln\left(\frac{RH}{100\%}\right) + \frac{m \cdot t}{T_n + t}}{m - \left[\ln\left(\frac{RH}{100\%}\right) + \frac{m \cdot t}{T_n + t}\right]}$$

$t_d$	dew point temperature in °C
$t$	actual temperature in °C
$RH$	actual relative humidity in %
$m$	17.62
$T_n$	243.12 °C

Sample code: `H = (log10(RH)-2.0)/0.4343+(17.62*t)/(243.12+t);`  
`td = 243.12*H/(17.62-H);`

### 3 Absolute humidity

Definition: The absolute humidity is the mass of water vapor in a particular volume of dry air. The unit is g/m<sup>3</sup>.

$$d_v(t, RH) = 216.7 \cdot \left[ \frac{\frac{RH}{100\%} \cdot A \cdot \exp\left(\frac{m \cdot t}{T_n + t}\right)}{273.15 + t} \right]$$

$d_v$	absolute humidity in g/m <sup>3</sup>
$t$	actual temperature in °C
$RH$	actual relative humidity in %
$m$	17.62
$T_n$	243.12 °C
$A$	6.112 hPa

Sample code: `dv = 216.7*(RH/100.0*6.112*exp(17.62*t/(243.12+t)))/(273.15+t);`

## 4 Mixing ratio

Definition: The mixing ratio is the mass of water vapor in a particular mass of dry air. The unit is g/kg.

$$r(t, RH) = \frac{622 \cdot \frac{RH}{100\%} \cdot A \cdot \exp\left(\frac{m \cdot t}{T_n + t}\right)}{p - \frac{RH}{100\%} \cdot A \cdot \exp\left(\frac{m \cdot t}{T_n + t}\right)}$$

*r* mixing ratio in g/kg  
*t* actual temperature in °C  
*RH* actual relative humidity in %  
*p* barometric air pressure in hPa  
*m* 17.62  
*T<sub>n</sub>* 243.12 °C  
*A* 6.112 hPa

Sample code: `e = RH/100.0*6.112*exp(17.62*t/(243.12+t));`  
`r = 622.0*e/(p-e);`

## 5 Heat index

Definition: The heat index is determined according to the National Weather Service and Weather Forecast Office of the National Oceanic and Atmospheric Administration (NOAA).

$$HI_{Celsius}(t, RH) = t + \frac{5}{9} \cdot \left[ \frac{RH}{100\%} \cdot \exp\left(\frac{m \cdot t}{T_n + t}\right) - 10 \right]$$

$$HI_{Fahrenheit}(t, RH) = \frac{9}{5} \cdot HI_{Celsius} + 32$$

*HI<sub>Celsius</sub>* Heat index in °C  
*HI<sub>Fahrenheit</sub>* Heat index in °F  
*t* actual temperature in °C  
*RH* actual relative humidity in %  
*m* 17.62  
*T<sub>n</sub>* 243.12 °C

Sample code: `p = RH/100.0*exp(17.62*t/(243.12+t));`  
`HIC = t+5.0/9.0*(p-10.0); // this is the heat index in Celsius`  
`HIF = 9.0/5.0*HIC+32.0; // this is the heat index in Fahrenheit`

### Revision History

Date	Revision	Changes
Aug. 20, 2008	1.0	Initial version

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